

Monthly Progressive Test (Solution)

Class: IX

A cademic
Excellence
Programme
TECHNO ACE

Subject: PCMB

Physics

$$W = mgh = (0.5) (9.8) (1)$$
$$= 4.9J$$

$$W = (200) (5) \cos 60^{\circ}$$

= 500 J.

W = change in kinetic energy

$$= \frac{1}{2} \times 0.25 \times (4)^2$$
$$= 2J$$

$$\frac{1}{2}mu^2 = \frac{1}{2}(1)(2)^2$$
$$= 2 J$$

W =
$$\frac{1}{2}$$
mu² = mgH
= $\frac{1}{2}$ ×(1)(10)² = 50 J (Against Gravity)
∴ W = -50 J

6. ®

Work done by friction can be positive, negative and zero

7. **(A)**

1 H.
$$P = 746$$
 Watt.

1 KWh = 1000 (Watt) × 3600 (s)
=
$$3.6 \times 10^6$$
 J

9. A

$$100 \text{ MW} = 100 \text{ M (J/s)}$$
 as $1 \text{ W} = 1 \text{ J/s}$.

$$\frac{25 \times 10 \times 1}{10} = 25 \text{ J/s}$$

11. B

A progressive wave is advancing a "disturbance" in elastic medium.

12. A

When wave passes through a medium, the particles of medium oscillate about mean position.

13. **©**

In case of longitudinal wave, oscillation of particles of medium is along the direction of propagation of wave.

14. B

As speed of sound in in moistair is greater than the speed of sound in dry air.

15. ©

As speed of sound in steel is greater than the speed of sound in water.

16. ©

As the range of audible sound is 20HZ to 20 KHZ.

17. A

Speed of sound = frequency \times wavelength

18. ®

As elastic medium is required for the sound to pass through.

19. A

$$V = f_1 \times \lambda_1 = f_2 \times \lambda_2$$
$$\frac{f_1}{f_2} = \frac{\lambda_2}{\lambda_1} = \frac{3}{2}$$

20. [®]

ultrasonic sound has frequency greater than 20 KHZ.

21. B

Both are true, but Reason is not correct explanation of assertion.

22. A

h =
$$\frac{1}{2}$$
gt² \Rightarrow 80 = $\left(\frac{1}{2}\right)$ (10)(t²) \Rightarrow t = 4s.
p = $\frac{\text{mgh}}{\text{t}}$ = $\frac{(1)(10)(80)}{4}$ = 200 watt.

23. ©

frequency = 200oscillations per second. = 200 HZ.

24. ©

$$T = \frac{15}{300} = \frac{1}{20} = 0.05s.$$

25. ®

frequency of source = frequency of sound wave = 100 HZ.

Chemistry

26. ®

An alpha (α) particle is a helium nucleus. i.e. $\alpha = {}^{4}\text{He}_{2}^{++}$

27. B

Formula of oxide is Y₂O₅

So, valency of Y = 5

Valency of Cl = 1

: Formula of the chloride of Y and Cl is YCl₅

28. ©

Formula of silver chromate is Ag₂CrO₄.

29. B

$$N_2 + 3H_2 \longrightarrow 2NH_3$$

28g 6g
$$2 \times 17 = 34g$$

$$34 \text{ g NH}_3 \equiv 28 \text{ g N}_2$$

$$6.8 \,\mathrm{g}$$
 ,, $\equiv 28 \times 6.8$

$$\frac{}{}$$
 g N_2

$$\equiv 5.6 \,\mathrm{g\,N_2}$$

$$\therefore H_2 = 6.8 - 5.6 = 1.2 g$$

30. A

1 a.m.u = $\frac{1}{12}$ th of mass of C-12 atom

31. **(A)**

Ratio by mass of the element present in $SO_2 = 1 \times 32 : 2 \times 16$

$$= 1:1$$

32. ©

M. W. of
$$O_2 = 2 \times 16 = 32$$

", CO₂ =
$$12 + 2 \times 16 = 44$$

", SO₂ =
$$32 + 2 \times 16 = 64$$

",
$$H_2S = 2 + 32 = 34$$

 \therefore M.W. of SO₂ = 2 × M.W. of O₂

33. [®]

$$x Mg + yN_2 \longrightarrow z Mg_3N_2$$

$$3Mg+1N_2 \longrightarrow 1Mg_3N_2$$

Comparing: x = 3, y = 1, z = 1

34. ®

Elements 'X' & 'Y' react to form X_aY_b

 \therefore Valency of 'X' = b

Elements 'P' & 'Q' react to form P_mQ_n.

∴ Valency of Q is m

 \therefore The formula of the compound formed by X & Q is X_mQ_b .

35. ®

M.W. of copper (II) sulphate = 160; M.W. of water = 18.

:. The % of water =
$$\frac{5 \times 18 \times 100}{(160 + 5 \times 18)}$$

36. A

Both assertion and reasons are correct and reason is the correct explanation of assertion. Thus the answer is A.

37. A

Both assertion and reasons are correct and reason is the correct explanation of assertion. Thus the answer is A.

38. ©

Assertion is correct but reason is incorrect. Thus the answer is C.

39. A

40. ©

1 G.M.W of
$$N_2 = 28$$
 g of $N_2 \equiv 1$ mole of N_2 molecule

$$\equiv 6.022 \times 10^{23}$$
 molecules

41. B

Ratio of number of moles of a component to the total number of moles is known a mole fraction. Molality (m') is the number of moles of solute per kilogram of solvent.

Molarity (M') is the number of moles of solution per litre of solution.

42. B

2, 8, 2.

$$Mg_{12}$$
; Proton (P) = Electron (d) = 12

 \therefore Electronic configuration of $Mg_{12} = K_2L_8M_2$

43. ©

Chronological order is II, I, III; where

I → Rutherford's atomic model

II → Thomson's atomic model

III → Bohr's atomic model

44. A

Mass of an electron is: 9.1083×10^{-3} kg.

A natural phenomenon that supports the experimental emulsion that atoms are divisible is radioactivity.

46. ®

X²⁻ contains 10 electrons & 8 Neutron.

So, atomic number = 10 - 2 = 8.

Atomic Mass = P + N = 8 + 8 = 16

47. A

 A^{3+} ; A_{13} has electronic configuration = $K_2L_8M_3$

Octet complete when it looses 3 electrons

$$A_{13} - 3 \longrightarrow A^{3+}$$
; (stable ion)

48. ©

10; A_{10} , number of protons = 10 = number of electron.

$$A_{10} - 2e^{\Theta} \rightarrow A_{10}^{2+}$$

So, number of protons A^{2+} is also 10.

49. ©

An atom of an element has 26 electrons and mass number = 56;

Mass number = P + N

: Neutron (N) =
$$56 - 26 = 30$$

50. ®

The presence of the three unpaired electrons in nitrogen atom is given by Hund's rule.

Mathematics

51. (A)

$$x^y = y^x \Longrightarrow y = x^{\frac{y}{x}}$$

$$\therefore \left(\frac{x}{y}\right)^{\frac{x}{y}} = \left(\frac{x}{\frac{y}{x^{\frac{x}{y}}}}\right)^{\frac{x}{y}} = \frac{\frac{x}{y}}{x} = x^{\frac{x}{y}-1}$$

52. B

$$108 \times 192 = 3^3 \times 2^2 \times 2^6 \times 3 = 3^4 \times 2^8$$

 \therefore Sum of powers of prime factors = 4 + 8 = 12

53. A

$$3\sqrt[3]{2} \times 7\sqrt[3]{6} \times 5\sqrt[3]{18}$$

$$= 105\sqrt[3]{2\times6\times18} = 105\sqrt[3]{6\times6\times6} = 105\times6 = 630$$

$$\sqrt{[0.04 \times 0.4 \times x]} = 0.4 \times 0.04 \times \sqrt{y}$$

$$\Rightarrow 0.04 \times 0.4 \times x = (0.4)^2 \times (0.04)^2 \times y$$

$$\Rightarrow x = 0.4 \times 0.04 \times y$$

$$\Rightarrow \frac{x}{y} = 0.016$$

55. ©

$$\frac{1}{11} = 0.090909 \dots = 0.\overline{09}$$

56. A

$$(3x + 2y)^2 = 144$$

 $\Rightarrow 9x^2 + 4y^2 + 12xy = 144$
 $\Rightarrow 9x^2 + 4y^2 + 72 = 144$
 $\Rightarrow 9x^2 + 4y^2 = 72$

57. ®

$$px^2 + qx + rx^4 + 5$$

Degree of this polynomial = 4
∴ It is biquadratic.

58. B

$$(x+2) \text{ is a factor of } x^3 - 2ax^2 + 16$$
∴
$$-8 - 8a + 16 = 0$$

$$\Rightarrow -8a = -8$$

$$\Rightarrow a = 1$$

59. ®

A quadratic polynomial can have at most 3 terms.

60. ©

$$x^{3} + 8y^{3} + z^{3} - 6xyz$$

$$= (x)^{3} + (2y)^{3} + (z)^{3} - 3 \times x \times 2y \times z$$

$$= (x + 2y + z)(x^{2} + 4y^{2} + z^{2} - 2xy - 2yz - xz)$$

61. ©

The co-ordinates of the point which lies on the y-axis and is 3 units away from the origin are (0, 3) and (0, -3).

62. [®]

If a point is equidistant from co-ordinate axes, then the line passing through this point and origin makes an angle 45° with the x-axis.

63. A

When the point (a, b) is reflected over the origin, then its image is (-a, -b).

The point (3, 4) lies on
$$y = mx + 8$$

 $\therefore 4 = 3m + 8 = 3m = -4$

$$\Rightarrow$$
 m = $\frac{-4}{3}$

65. ©

The common point of lines x = 0 and y = 0 is (0, 0) only. So, number of common points = 1

66. ©

The total cost = ₹
$$(50x + 30y)$$

∴ $50x + 30y = 200$
⇒ $5x + 3y = 20$.

67. ®

$$5x + 3 \times 2.5 = 20$$

$$\Rightarrow 5x = 20 - 7.5 = 12.5$$

$$\Rightarrow x = 2.5$$

68. ®

$$5x + 3y = 20$$

For $x = 0$, $y = \frac{20}{3}$

$$\therefore$$
 The graph of $5x + 3y = 20$ cuts $y - axis$ at $\left(0, \frac{20}{3}\right)$

69. ®

$$p(x) = x^2 - 4x + 3 = (x - 3)(x - 1)$$

∴ Zeroes are 1 and 3

∴ Assertion (A) is true.

Number of zeroes of a polynomial cannot exceed its degree. Reason (R) is also correct. But reason is not the correct explanation of Assertion (A).

70. ©

Sum of
$$2 + \sqrt{3}$$
 and $4 + \sqrt{3} = 6 + 2\sqrt{3}$ which is irrational.

∴ Assertion (A) is true

Sum of two irrational numbers is not always irrational number.

For example, $3 + \sqrt{5} + 3 - \sqrt{5} = 6$ which is rational.

∴ Reason is false.

$$\sqrt[4]{x} + \frac{1}{\sqrt[4]{x}} = 2$$

$$\Rightarrow \sqrt{x} + \frac{1}{\sqrt{x}} = 2$$

$$\Rightarrow x + \frac{1}{x} = 2$$

$$\Rightarrow x^2 + 1 = 2x$$

$$\Rightarrow (x - 1)^2 = 0$$

$$\Rightarrow x = 1$$

$$\therefore x^{2024} + \frac{1}{x^{2024}} = 1 + \frac{1}{1} = 2$$

$$x^{2023} + 2 = (x^2 - 1) \times g(x) + ax + b$$

For
$$x = 1$$

$$3 = a + b$$

For
$$x = -1$$

$$1 = -a + b$$

$$\therefore 2b \Longrightarrow 4 = b = 2$$

$$\therefore$$
 a = 1

$$\therefore$$
 Remainder = $x + 2$

73. ©

Name of quadrilateral ABCD is square because all sides are equal and also diagonals are equal.

74. ©

The number of solution for 2x + 3 = 0 is only one in one dimension. Again, the number of solutions for 2x + 3 = 0 is infinitely many in two dimensions.

75. ©

$$(x+1)^{2} + (x+2)^{2} + (x+3)^{2} - (x+1)(x+2) - (x+2)(x+3) - (x+3)(x+1)$$

$$= \frac{1}{2} [2a^{2} + 2b^{2} + 2c^{2} - 2ab - 2bc - 2ca] \text{ where } a = x+1, \ b = x+2, \ c = x+3$$

$$= \frac{1}{2} [(a-b)^{2} + (b-c)^{2} + (c-a)^{2}]$$

$$= \frac{1}{2} [1+1+4] = 3$$

76. B

Muscle cell

77. A

Cell sap

78. A

2

Meristematic and permanent tissues

79. B

Slower than myelinated neuron

80. B

Collenchyma

81. ®

Cellulose + pectin + hemicellulose

82. A

Roots

83. ®

Tensile strength

Flexibility, i.e, ability to bend without breaking

84. ®

Simple permanent tissue

Tissue made of the same type of cells. Cells generally do not divide and are specialized functionally

85. A

Thin and flat

86. ©

Basement membrane

Characteristic feature of epithelial tissues

87. B

Cuboidal epithelium

88. ©

Testis

It produces the gametes

89. B

Squamous epithelium of skin

90. A

Single membrane covering of vacuoles

91. B

Tendons connect cartilage to bones Tendons connect muscles to bones

92. [©]

Collenchyma

It is a simple permanent tissue not found in the xylem

93. A

Both A and R are true and R is the correct explanation of A

94. ©

A is true but R is false

Lysosomes are organelles present in the cytoplasm

95. B

Both A and R are true but R is not the correct explanation of A

96. ©

Intercropping

Each type of plant is assigned a separate row

97. A

Protein

98. ®

All of the above

99. A

Cereal crop

100. A

Cereal crop