



TECHNO INDIA GROUP PUBLIC SCHOOL

JEE Mock Test Series-II (Part-2)

SOLUTION

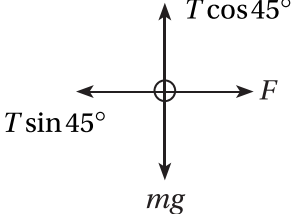
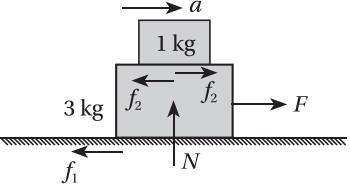
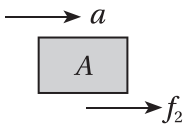
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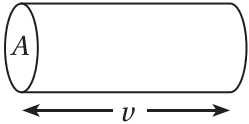
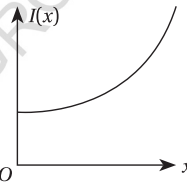
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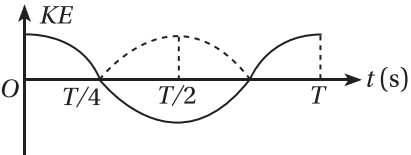
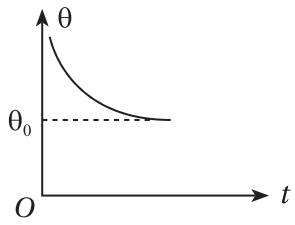
PHYSICS

SECTION A

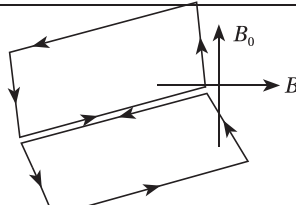

Section A consists of 20 questions of 4 mark each.

1.	<p>①</p> $m = \frac{4}{3}\pi R^3 \times \rho \Rightarrow \ln m = \ln(4\pi/3) + 3\ln R + \ln \rho$ $0 = 0 + 3\frac{dR}{Rdt} + \frac{1}{\rho} \frac{d\rho}{dt}$ $\frac{dR}{dt} = v \therefore v \propto R$	
2.	<p>①</p> $ \vec{v} = a\omega \sqrt{(-\sin \omega t)^2 + (\cos \omega t)^2 + 1} = \sqrt{2}a\omega$	
3.	<p>③</p>  <p style="text-align: right;">$\therefore F = mg = (10)(10) = 100 \text{ N}$</p>	
4.	<p>②</p>   $F - f_1 = (m_A + m_B)a \quad \dots (1) \quad f_2 = m_A \cdot a = (0.2)(1)(10)$ $a = \mu g = (0.2)(10) = 2 \text{ m/s}^2 \quad \dots (2)$ $f_1 = \mu N = (0.2)(1 + 3)(10) = 8 \text{ N} \quad \dots (3)$ $\therefore F - 8 = 8 \quad F = 16 \text{ N}$	

5.	<p>④</p> $\theta = \omega t = \omega \frac{\pi}{2\omega} = \left(\frac{\pi}{2}\right)$ <p>at t (s): </p> $V_A = \omega R_1 (-\hat{i})$ $V_B = \omega R_2 (-\hat{i})$ <p></p> V_B $V_A - V_B = -\omega R_1 \hat{i} + \omega R_2 \hat{i} = \omega(R_2 - R_1) \hat{i}$		
6.	<p>③</p> $P = Fv \quad F = v \frac{dm}{dt} = v \frac{d(\rho \times V)}{dt} = \rho v \frac{d(\text{Volume})}{dt}$ $= \rho v Av = \rho Av^2$ $\therefore P = (\rho Av^2)v = \rho Av^3 \quad \therefore P \propto v^3$		
7.	<p>④</p> $20 = v_1 \times \sqrt{\frac{2h}{g}} = v_1 \times \sqrt{\frac{2 \times 5}{10}} = v_1$ $100 = v_2 \times \sqrt{\frac{2h}{g}} = v_2$ $0.01 \times v = 0.01 \times 100 + 0.2 \times 20 = 5 \quad v = 500 \text{ m/s}$		
8.	<p>②</p> $I(x) = \frac{2}{5} MR^2 + Mx^2$		
9.	<p>①</p> $v_i = -\frac{GM}{R^3} (1.5R^2 - 0.5r^2)$ $v_p = -\frac{GM}{R^3} \left[1.5R^2 - 0.5 \times \left(\frac{R}{2}\right)^2 \right] = -\frac{GM}{R} \left(1.5 - 0.5 \times \frac{1}{4} \right) = -\frac{GM}{R} \left(\frac{6-0.5}{4} \right) = -GM \times \frac{5.5}{4R}$ $v'_p = v'_c = -\frac{3GM}{8R}$ $v_R = v_p - v'_p = -\frac{GM}{R} \times \frac{5.5}{4} + \frac{3GM}{8R} = \frac{-11GM + 3GM}{8R} = -\frac{8GM}{8R} = -\frac{GM}{R}$		

10.	<p>③</p> $y = A \sin \omega t$ $\frac{dy}{dt} = A \omega \cos \omega t = v$ $k = \frac{1}{2} m v^2 = \frac{1}{2} m A^2 \omega^2 \cdot \cos^2 \omega t$ 	
12.	<p>①</p> $v_T \propto r^2 \quad \therefore \frac{v_1}{v_2} = \frac{R^2}{(R/3)^2} = 9$	
11.	<p>①</p> $\frac{4}{3} \pi R^3 \rho g + kx = \frac{4}{3} \pi R^3 \cdot 2 \rho \cdot g \quad \dots (1)$ $\frac{4}{3} \pi R^3 \cdot 3 \rho g - kx = \frac{4}{3} \pi R^3 \cdot 2 \rho g \quad \dots (2)$ $kx = \frac{4}{3} \pi R^3 \cdot \rho g$ $x = \left(\frac{4 \pi R^3}{3k} \right) \rho g$	
13.	<p>①</p> $L = 10 \log_{10} \frac{I}{I_0} = 120 \Rightarrow 10^{12} = \frac{I}{10^{-12}}$ $\therefore I = 1 \text{ w/m}^2 = \frac{\rho}{4\pi r^2} \Rightarrow r = \sqrt{\frac{\rho}{4\pi}} = \sqrt{\frac{2}{4\pi}} = \frac{1}{\sqrt{2\pi}} \approx 0.4 \text{ m} \approx 40 \text{ cm}$	
14.	<p>①</p> $W = nR(T_2 - T_1) = 0.5 \times 8.31 \times 70 = 8.31 \times 35 = 291 \text{ J}$	
15.	<p>③</p> $-ms \frac{d\theta}{dt} = k(\theta - \theta_0) \Rightarrow \int \frac{d\theta}{\theta - \theta_0} = -\left(\frac{k}{ms} \right) \int dt$ $\ln(\theta - \theta_0) = -\frac{kt}{ms} + c; \quad t = 0; \theta = \theta_i; c = \ln(\theta_i - \theta_0)$ $\frac{\theta - \theta_0}{\theta_i - \theta_0} = e^{-\lambda t} \Rightarrow \theta = \theta_0 + e^{\lambda t} (\theta_i - \theta_0) \quad \theta = \theta_0 + (\theta_i - \theta_0) e^{-\lambda t}$ 	

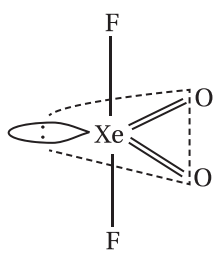
<p>16. ①</p>	<p> $\sin \alpha = \sqrt{2} \cdot \sin r_1$ $\Rightarrow \frac{1}{\sqrt{2}} = \sqrt{2} \sin r_1 \Rightarrow r_1 = 30^\circ$ $\sin \theta_c = \frac{1}{n} = \frac{1}{\sqrt{2}} \Rightarrow \theta_c = 45^\circ$ $90^\circ + r_1 + 90^\circ - \theta_c + \theta = 180^\circ$ $\therefore \theta = \theta_c - r_1 = 15^\circ$ </p>	
<p>17. ②</p>	<p> $\frac{\mu_2 - \mu_1}{v_1} = \frac{\mu_2 - \mu_1}{\alpha + R}$ </p> <p> $\frac{\mu_3 - \mu_2}{v_2} = \frac{\mu_3 - \mu_2}{\alpha + R}$ </p> <p> <u>Add</u> $\frac{\mu_3}{v_2} = \frac{\mu_3 - \mu_1}{R}$ $v_2 = f \Rightarrow \left(\frac{\mu_3}{\mu_3 - \mu_1} \right) R$ </p>	
<p>18. ①</p>	<p> $\Delta x (\text{extra}) = (\mu t - t) = n\lambda$ $(\mu - 1)t = n\lambda$ Put $n = 1$ $(\mu - 1)t = \lambda$ $\mu = 1.5$ $t_{\min} = 2\lambda$ </p>	
<p>19. ①</p>	<p> $\frac{3R}{4} + \frac{R}{8} = \frac{7R}{8}$ $\frac{7R}{8} \parallel \frac{R}{8} = \frac{\frac{7R}{8} \times \frac{R}{8}}{R} = \frac{7}{64} R$ </p>	

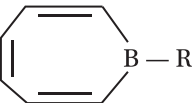
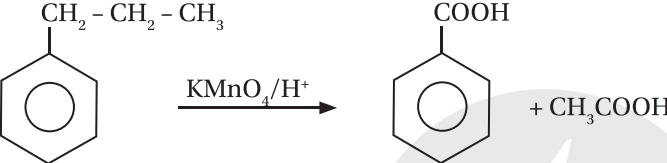
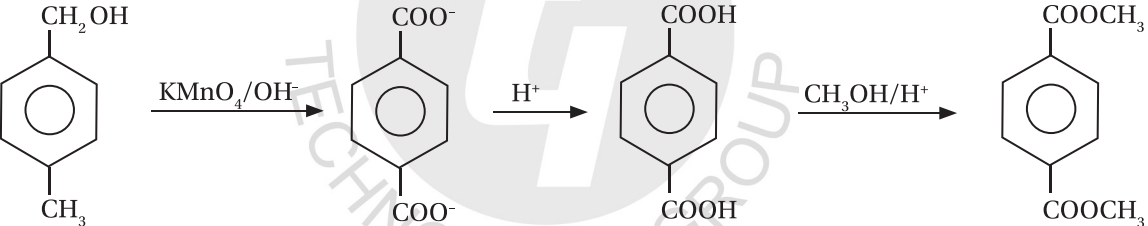
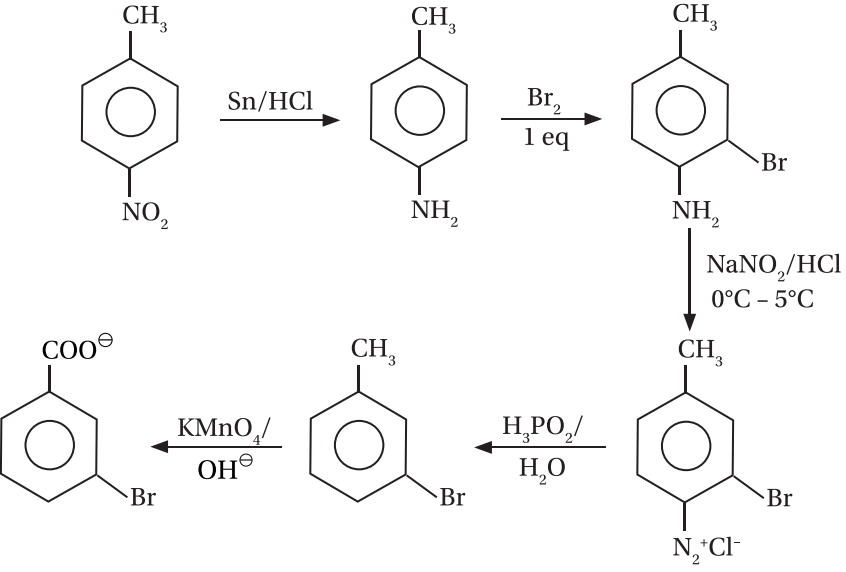
20.	<p>①</p> $Q_2 = \frac{2cE}{c+3} = \frac{2E}{1+(3/c)} \quad c \uparrow \text{ then } Q_2 (\uparrow) \text{ non-linearly}$ $\frac{d(Q_2)}{dc} = \frac{6E}{(c+3)^2} \quad \text{+ve slope decreases with } c \uparrow$	
<div style="background-color: black; color: white; padding: 2px 10px; display: inline-block; margin-bottom: 5px;">SECTION B</div> <p>Section B consists of 5 questions of 4 marks each.</p>		
21.	$(18)(5) \cos 0^\circ - (1)(10)(4) = k_f - 0 \Rightarrow 90 - 40 = k_f = 5 \times 10 \quad \text{Ans: } n = 5$	
22.	$\vec{B} = B_0 \hat{i} + B_0 \hat{k}$ $\hat{B} = \frac{\hat{i} + \hat{k}}{\sqrt{2}}$ $n = \sqrt{2} \quad n^2 = 2$ 	
23.	$i = i_0 \left(1 - e^{-\frac{Rt}{L}} \right) \quad i^2 R = \frac{d}{dt} \left(\frac{1}{2} Li^2 \right) \Rightarrow iR = L \frac{di}{dt} \quad i_0 R (1 - e^{-Rt/L}) = L \cdot i_0 \cdot \frac{R}{L} e^{-Rt/L}$ $1 - e^{-Rt/L} = e^{-Rt/L} \Rightarrow 2e^{-Rt/L} = 1 \quad \therefore \frac{t}{\ln 2} = (2) \text{ Ans.}$	
24.	 $mv = mv_A + \frac{m}{2}v_B$ $2v = 2v_A + v_B \quad \dots (1)$ $e = \frac{v_B - v_A}{v} = 1 \quad \therefore v_B - v_A = v \quad \dots (2)$ $v_B = 4v/3$ $v_A = v/3 \quad \frac{\lambda_A}{\lambda_B} = \frac{P_B}{P_A} = \frac{\frac{m}{2} \cdot 4v/3}{mv} = 2 \quad \left[\because \lambda = \frac{h}{P} \right]$	
25.	<p>Step I : Constituent proton and neutron combined mass > mass of Nucleus. This difference is called mass defect which is responsible for mass defect.</p> $\text{mass defect}_1 = 10(m_p + m_n) - M_1 \quad \text{mass defect}_2 = 20(m_p + m_n) - M_2$ <p>heavier the nucleus, more is the mass defect</p> $\therefore 20(m_p + m_n) m_2 > 10(m_p + m_n) - M_1$ $\Rightarrow 10(m_p + m_n) > M_2 - M_1 \quad \Rightarrow M_2 < M_1 + 10(m_p + m_n)$ <p>but $M_1 < 10(m_p + m_n) \quad M_2 < 20(m_p + m_n)$</p> $\therefore M_2 < 2M_1 \Rightarrow \frac{M_2}{M_1} < 2$	

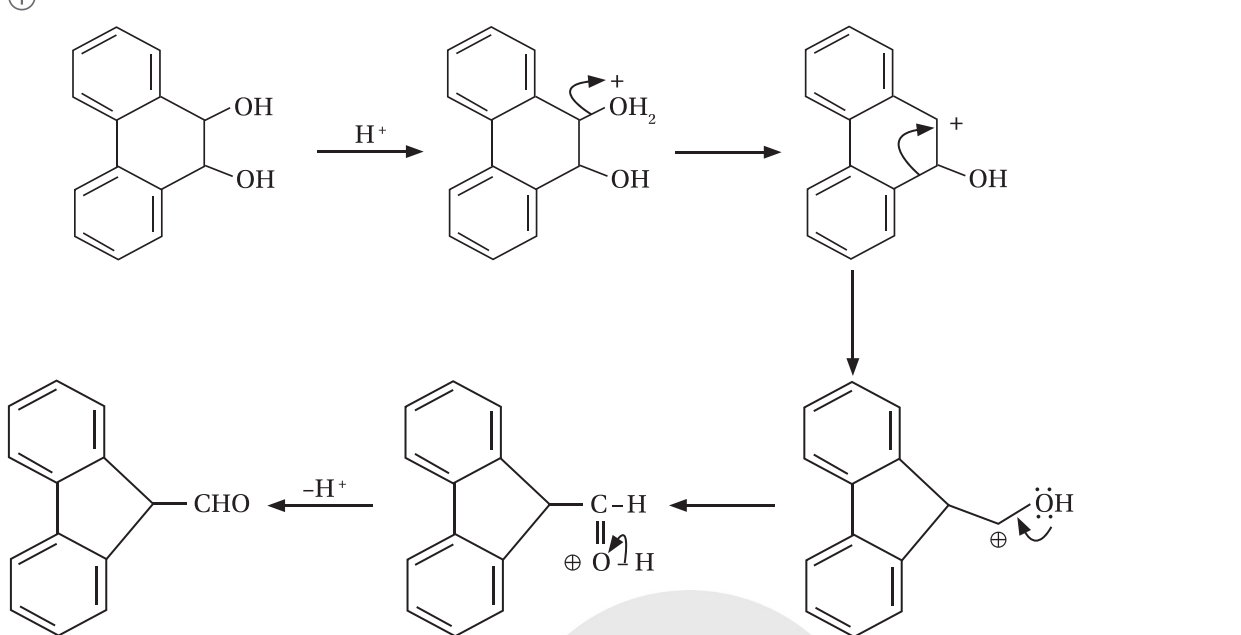
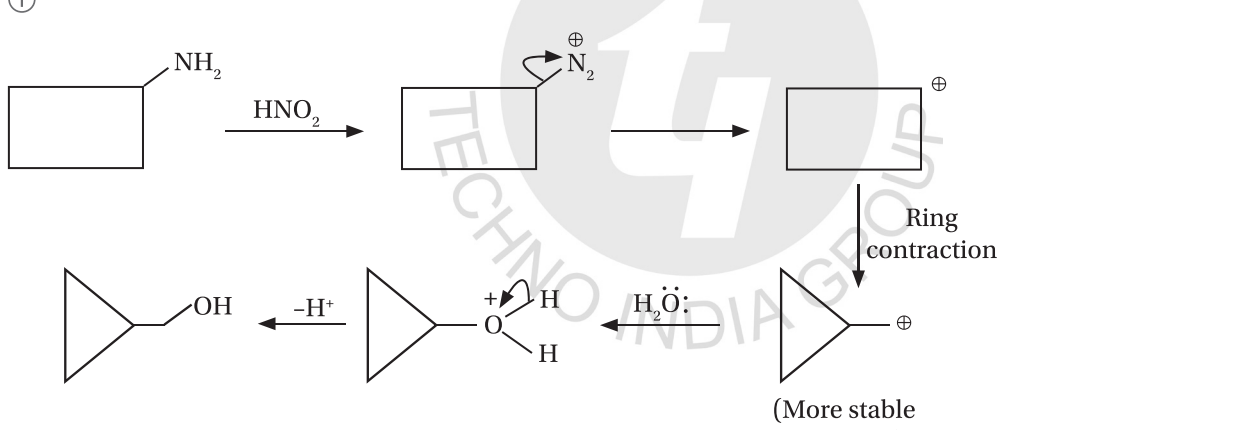
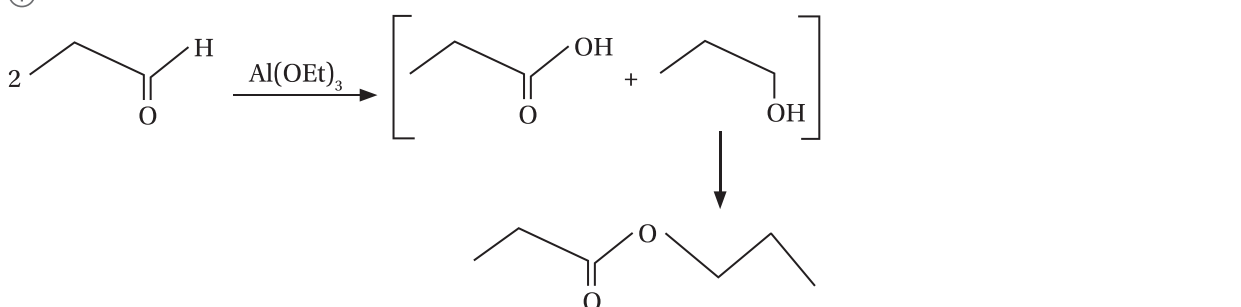
CHEMISTRY**SECTION A**

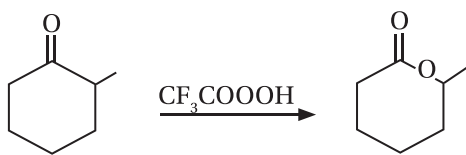
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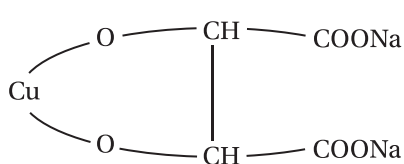
26.	<p>③</p> $\mu = \sqrt{n(n+2)} = \sqrt{35}$ $\Rightarrow n^2 + 2n - 35 = 0$ $\Rightarrow n^2 + 7n - 5n - 35 = 0$ $\Rightarrow (n+7) - 5(n+7) = 0$ $\Rightarrow (n+7)(n-5) = 0$ $\therefore n = 5$ <p>But, M^{3+} ion; so, total electrons = $5 - 3 = 2$.</p>
27.	<p>③</p> $3P \Rightarrow n = 3, l = 1$ $\therefore \text{Radial nodes} = n - l - 1 = 3 - 1 - 1 = 1$
28.	<p>①</p> <p>Normally, $N = n - f \times \text{Molarity (M)}$</p> $= 2 \times 0.2$ $= 0.4$ <p>100 ml 0.2 (M) H_2SO_4</p> $\equiv 100 \times 0.2 \text{ ml } 1(M) H_2SO_4$ $\equiv 20 \text{ ml } 1(M) H_2SO_4$ $\equiv 20 \times 2(N) H_2SO_4$ $\equiv 40 \text{ ml } 1(N) H_2SO_4$ <p>100 ml 0.2 (M) NaOH</p> $\equiv 100 \text{ ml} \times 0.2 \text{ ml } 1(M) \text{ NaOH}$ $\equiv 20 \text{ ml } 1(N) \text{ NaOH}$ $\equiv 20 \text{ ml } 1(N) H_2SO_4$
	<p>$\therefore \text{Excess} = 40 - 20 = 20 \text{ ml } 1(N) H_2SO_4$</p> <p>$\therefore \text{Let the strength of final volume} = x$</p> $\therefore 20 \times 1 = x \times 200$ $\Rightarrow x = \frac{20}{200} = 0.1$
29.	<p>④</p> $m = \frac{X_A \times 1000}{X_B \times M_B} = \frac{0.2 \times 1000}{0.8 \times 18}$ $= \frac{250}{18} = 13.8$

30.	<p>①</p> <p>CO : Total e's = 6 + 8 = 14 ; B . O = 3</p> <p>NO⁻ : ,, ,, = 7 + 8 + 1 = 16; B . O = 2.5</p> <p>NO⁺ : ,, ,, = 15 - 1 = 14, B . O = 3</p> <p>CN⁻ : ,, ,, = 14 ; B . O = 3</p> <p>N₂ : ,, ,, = 14 ; B . O = 3</p>										
31.	<p>④</p> <p>XeO₂F₂ ; H = $\frac{1}{2} (V + M - C + A)$</p> <p style="margin-left: 40px;">$= \frac{1}{2} (8 + 2) = 5 = sp^3d$</p> <p>l . p = 1</p> <p>So, according to V.S.E.P.R theory, sea-saw shape:</p> <p>According to Bent rule, more e.n. atom will occupy at axial position.</p> <div style="text-align: right;">  </div>										
32.	<p>④</p> $CH_3COOH + NaOH \longrightarrow CH_3COONa + H_2O$ <table style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 20%;">Initial m mole:</td> <td style="width: 20%; text-align: center;">5</td> <td style="width: 20%; text-align: center;">10</td> <td style="width: 20%; text-align: center;">0</td> <td style="width: 20%; text-align: center;">0</td> </tr> <tr> <td>m. mole after reaction:</td> <td style="text-align: center;">0</td> <td style="text-align: center;">10-5=5</td> <td style="text-align: center;">5</td> <td style="text-align: center;">5</td> </tr> </table> <p>Hydrolysis of CH₃COONa is suppressed by strong Base NaOH</p> $[OH^-] = \frac{5}{150} = \frac{1}{30}$ $pOH = -\log_{10}[OH^-] = -\log_{10}\left(\frac{1}{30}\right) = -\log 30 = 1.4771$ <p>pH + pOH = 14 ; pH = 14 - 1.4771 = 12.52</p>	Initial m mole:	5	10	0	0	m. mole after reaction:	0	10-5=5	5	5
Initial m mole:	5	10	0	0							
m. mole after reaction:	0	10-5=5	5	5							
33.	<p>②</p> $AgCl(s) \rightleftharpoons Ag^+(aq) + Cl^-(aq)$ $AgNO_3 \rightleftharpoons Ag^+ + NO_3^-$ <p style="margin-left: 40px;">0.1(M) 0.1 M</p> <p>$K_{sp} = [Ag^+]_{Total} [Cl^-] = (S + 0.1) S = S^2 + 0.1S$</p> <p>$\approx 0.1S$ [$\because S \ll 0.1$]</p> $\therefore S = \frac{K_{sp}}{0.1} = \frac{2.8 \times 10^{-10}}{0.1} = 2.8 \times 10^{-9} \text{ mol / k}$										
34.	<p>③</p> <p>1 M of 10 ml H₂SO₄ \equiv 1 M of 20 ml of NH₃</p> <p>1000 ml of 1(M) ammonia contains = 14 g N</p> <p>\therefore 20 ml of 1(M) ammonia contains = $\frac{14 \times 20}{1000}$ gN</p> <p>\therefore % of N = $\frac{14 \times 20}{1000 \times 0.5} \times 100 = 56.0\%$</p>										

35.	<p>③</p>  <p>Obeys $(4n + 2) \pi$ electrons i.e. Huckel's rule $n = 1$, $6\pi e^-$'s</p>
36.	<p>④</p> <p>Due to +R effect of $-\text{:OCH}_3$, electron density over 'N' atom in $-\text{NH}_2$ group further increased. Hence, it is most basic.</p>
37.	<p>①</p> 
38.	<p>④</p> 
39.	<p>②</p> 

40.	<p>①</p> 
41.	<p>①</p> 
42.	<p>②</p> <p>Fehling's solution: PhCHO does not give Fehling's solution test while aliphatic aldehyde give this test. All aldehydes give Tollen's reagent test.</p>
43.	<p>①</p> 

44.	③ Glucose does not give D.N.P test. (No free -CHO gr.)
45.	④ Baeyer's villiger oxidation: 
SECTION B Section B consists of 5 questions of 4 marks each.	
46.	(4) Equivalent weight of copper = $\frac{63.5}{2}$ \therefore 2 mol require 4F or 1 mole $\text{Cu}^{++} \equiv 63.5 \text{ g Cu}^{++}$ $\frac{63.5}{2} \text{ g Cu} \equiv 1 \text{ g equivalent weight of Cu}^{++}$ $\therefore 63.5 \text{ g of Cu}^{++} \equiv 2 \text{ g equivalent weight of Cu}^{++}$ $\therefore 1 \text{ mole Cu}^{++} \equiv 2 \text{ g equivalent weight of Cu}^{++}$ $2 \text{ mole Cu}^{++} \equiv 4 \text{ g equivalent weight of Cu}^{++} \equiv 4\text{F}$
47.	(3) $\text{K}_4[\text{Fe}(\text{CN})_6] \rightleftharpoons 4\text{K}^{\oplus} + [\text{Fe}(\text{CN})_6]^{4-}$ $x = 4 + 1 = 5$ $\alpha = \frac{1-i}{1-x} \Rightarrow 0.50 = \frac{1-i}{1-5} = \frac{1-i}{-4}$ $\Rightarrow 1 - i = -2.0$ $\Rightarrow i = 2 + 1 = 3$
48.	(0) $\frac{t_2}{t_1} = \left(\frac{a_1}{a_2}\right)^{n-1}$ $\Rightarrow \frac{160}{320} = \left(\frac{58}{29}\right)^{n-1}$ $\Rightarrow \left(\frac{1}{2}\right) = (2)^{n-1}$ $\Rightarrow 2^{-1} = 2^{n-1}$ $\therefore n - 1 = -1$ $\Rightarrow n = -1 + 1 = 0$

49.	(+6) $4\text{KOH} + 2\text{MnO}_2 + \text{O}_2 \longrightarrow 2\text{K}_2\text{MnO}_4 + 2\text{H}_2\text{O}$ (Potassium Manganate) K_2MnO_4 $2(+1) + x - 8 = 0$ $\Rightarrow 2 + x - 8 = 0$ $\Rightarrow x = +6.$
50.	(2) Fehling's solution is a complex of Cu^{2+} . Formula of Fehling solution is:  E.C. of $\text{Cu}^{2+} = 3d^9$ Number of unpaired electron in Cu^{2+} ion = 1 $\therefore \mu = \sqrt{n(n+2)} = \sqrt{1(1+2)} = 1.73 \approx 2\text{BM}$

Mathematics

SECTION A

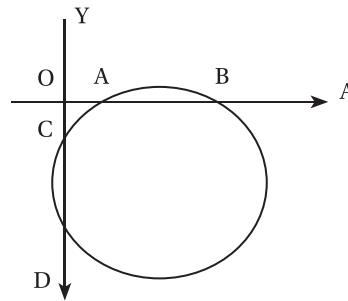
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	Solution	Answer	[4]
51.	$f(x) = \frac{\tan x}{\tan 3x} = \frac{1 - 3 \tan^2 x}{3 - \tan^2 x} \Rightarrow \tan^2 x = \frac{1 - 3f(x)}{3 - f(x)} \geq 0$ $\Rightarrow \left(f(x) \leq \frac{1}{3} \right) \cup (f(x) > 3)$	Ⓐ	[4]
52.	$x = \frac{50 + 59 + 46 + 62 + 48 + x}{6} \Rightarrow x = 53$ $\sigma^2 = \frac{\sum x_i^2}{n} - (\bar{x})^2 = \frac{100}{3} \Rightarrow \sigma = \frac{10}{\sqrt{3}}$	Ⓒ	[4]
53.	Let $P\left(Ct_1, \frac{C}{t_1}\right), Q\left(Ct_2, \frac{C}{t_2}\right)$, be the points. Circle : $(x - Ct_1)(x - Ct_2) + \left(y - \frac{C}{t_1}\right)\left(y - \frac{C}{t_2}\right) = 0$ (1) $m_{PQ} = -\frac{1}{t_1 t_2} = 1 \Rightarrow m_{PQ} = m_{y=x} \Rightarrow t_1 t_2 = -1$ (2)	Ⓐ	[4]

	<p>From (1) & (2) : $(x^2 + y^2 - 2c^2) - (t_1 + t_2)(Cx - Cy) = 0$ $C + \lambda L = 0$ Family of circle passing through the intersection of $C = 0$, and $L = 0$ $C : x^2 + y^2 - 2c^2 = 0$ $L : cx - cy = 0 \Rightarrow x = y \Rightarrow x = \pm c = y$ Points are (C, C) and $(-C, -C)$</p>		
54.	<p>$(x + a)(x + b)(x + c) \dots \dots \dots 50 \text{ terms} = x^{50} + x^{49} \cdot S_1 + x^{48} S_2 + \dots$ The Coefficient of x^{49} $S_1 = \left[\frac{C_1}{C_0} + 2^2 \cdot \frac{C_2}{C_1} + 3^2 \cdot \frac{C_3}{C_2} + \dots + r^2 \cdot \frac{C_r}{C_{r-1}} \right]$ $r^2 \cdot \frac{C_r}{C_{r-1}} = r^2 \cdot \frac{n-r+1}{r} = (51-r) = 51r - r^2 \quad (\because n = 50)$ $\sum_{r=1}^{50} r^2 \cdot \frac{C_r}{C_{r-1}} = 51 \sum_{r=1}^{50} r - \sum_{r=1}^{50} r^2 = 25 \times 17 \times 52$</p>	Ⓒ	[4]
55.	<p>For $n = 1$, $P(1) : (65 + k)$ is divisible by 64 $\therefore k$ should be -1</p>	Ⓐ	[4]
56.	<p>OR = 5, $C = -5$ $\frac{-D}{4a} = -9$ $\Rightarrow \frac{b^2 + 20a}{4a} = 9 \Rightarrow b^2 = 16a$ $Y = ax^2 + bx - 5$ $\text{ar}(\Delta OQB) = \frac{45}{4} \Rightarrow OB = \frac{5}{2}$ at $x = \frac{5}{2}$ $y = 0 \Rightarrow \frac{25a}{4} + \frac{5b}{2} - 5 = 0$ $\Rightarrow 5a + 2b - 4 = 0$ $\Rightarrow 5b^2 + 32b - 64 = 0 \quad (\because b^2 = 16a)$ $\Rightarrow b = -8, \frac{8}{5}$ $OB > OA \Rightarrow b = -8 \text{ \& } a = 4$ $Y = 4x^2 - 8x - 5 = (2x + 1)(2x - 5)$ For A $x = -\frac{1}{2} \therefore AB = 3$</p>	Ⓐ	[4]
57.	<p>Tangent : $y = mx \pm \sqrt{1+m^2}$ it passes through $(-2, 0)$ $m = \pm \frac{1}{\sqrt{3}}$ Equation of tangents are : $\sqrt{3}y = x + 2$ & $\sqrt{3}y = -x - 2$ Centre of smaller circle $(h_2, 0)$ $-1 - h_2 = h_2 + 2 \Rightarrow h_2 = -\frac{4}{3}$ Centre $\left(-\frac{4}{3}, 0\right)$, radius = $\frac{1}{3}$</p>	Ⓐ	[4]

58.	$\lim_{x \rightarrow 0} \frac{(2 - \sqrt{x+4})(2 + \sqrt{x+4})}{\sin 2x(2 + \sqrt{x+4})} = -\frac{1}{8}$	Ⓐ	[4]
59.	$S_{\infty} = \frac{a}{1-r} = 162, S_n = \frac{a(1-r^n)}{(1-r)} = 160$ <p style="text-align: center;"> $\xrightarrow{\quad} \boxed{1} \qquad \xrightarrow{\quad} \boxed{2}$ </p> $\boxed{1} \cap \boxed{2}: r^n = \frac{1}{81} = 3^{-4} \Rightarrow \left(\frac{1}{r}\right)^n = 81$ $\left. \begin{array}{l} \frac{1}{r} = 3, 9, 81 \\ n = 4, 2, 1 \end{array} \right\}$ $a = 162 \left(1 - \frac{1}{3}\right) = 108$ $a = 162 \left(1 - \frac{1}{9}\right) = 144$ $a = 162 \left(1 - \frac{1}{81}\right) = 160$ $\text{Sum} = 108 + 144 + 160 = 412.$	Ⓒ	[4]
60.	$\lim_{x \rightarrow 0} \frac{(-2 \sin^2 \frac{x}{2}) \left[\left(1 - \frac{x^2}{2!} + \frac{x^4}{4!} - \dots\right) - \left(1 + x + \frac{x^2}{2!} + \frac{x^3}{3!} + \dots\right) \right]}{x^n}$ $= \lim_{x \rightarrow 0} (-2) \cdot \left(\frac{\sin(\frac{x}{2})}{\frac{x}{2}}\right)^2 \cdot \frac{1}{4} \cdot \frac{\left(-x - \frac{2x^2}{2!} + \frac{x^3}{3!} + \dots\right)}{x^{n-2}}$ $= \lim_{x \rightarrow 0} \left(-\frac{1}{2}\right) \frac{\left(-1 - \frac{2x}{2!} - \frac{x^3}{3!} + \dots\right)}{x^{n-3}} = \text{a finite}$ $n = 3 \Rightarrow \text{limit} = \frac{1}{2}$	Ⓒ	[4]
61.	<p>For non trivial solution,</p> $\begin{vmatrix} p+a & b & c \\ a & q+b & c \\ a & b & r+c \end{vmatrix} = 0$ $\Rightarrow \begin{vmatrix} p & -q & 0 \\ 0 & q & -r \\ a & b & r+c \end{vmatrix} = 0 \quad \begin{array}{l} (R_1 \rightarrow R_1 - R_2 \\ R_2 \rightarrow R_2 - R_3) \end{array}$ $\Rightarrow pqr + pqc + prb + qra = 0$ $\Rightarrow \frac{a}{p} + \frac{b}{q} + \frac{c}{r} = -1$	Ⓐ	[4]
62.	B is symmetric but it is not transitive as $(r, p), (p, q) \in B$ but $(r, q) \notin B$	Ⓒ	[4]

63.	$32 \sin \frac{A}{2} \cdot \sin \frac{SA}{2}$ $= 2 \cdot 16 \sin \frac{A}{2} \cdot \sin \frac{SA}{2}$ $= 16 (\cos 2A - \cos 3A)$ $= 16 (2 \cos^2 A - 1 - 4 \cos^3 A + 3 \cos A) = 16 \left(2 \cdot \left(\frac{3}{4} \right)^2 - 1 - 4 \cdot \left(\frac{3}{4} \right)^3 + 3 \cdot \left(\frac{3}{4} \right) \right) = 11.$	Ⓐ	[4]
64.	<p>For concurrency,</p> $\begin{vmatrix} a & b & c \\ b & c & a \\ c & a & b \end{vmatrix} = 0.$ $\Rightarrow a^3 + b^3 + c^3 - 3abc = 0.$	Ⓒ	[4]
65.	$\frac{dy}{dx} = \frac{y^2 - x^2}{2xy} \text{ put } y = vx. \Rightarrow \int \frac{2v}{v^2 + 1} dv = - \int \frac{dx}{x}$ $\Rightarrow x \left(\frac{y^2}{x^2} + 1 \right) = c$ <p>When $x = 1$, then $y = 1 \Rightarrow c = 2$.</p> <p>Curve : $x^2 + y^2 - 2x = 0$.</p>	Ⓐ	[4]
66.	$f(x) = -f(-x) \Rightarrow f(x) + f(-x) = 0.$ $\Rightarrow \frac{x^2 + 5}{\lambda} - \left\{ \frac{x^2 + 5}{\lambda} \right\} = 0$ $\Rightarrow \left[\frac{x^2 + 5}{\lambda} \right] = 0$ $\Rightarrow 0 \leq \frac{x^2 + 5}{\lambda} < 1$ <p>As $x \in [-3, 3] \Rightarrow x^2 + 5 \in [5, 14] \Rightarrow \lambda > 14$.</p>	Ⓒ	[4]
67.	<p>$h = 0$ for circle.</p> $AB > 0 \Rightarrow g^2 - c > 0 \Rightarrow g^2 > c$ $CD > 0 \Rightarrow f^2 - c > 0 \Rightarrow f^2 > c$ <p>As origin lies outside the circle</p> $\therefore c > 0.$ <p>Required conditions : $g^2 > c, f^2 > c, c > 0$</p> $h = 0.$	Ⓒ	[4]
68.	$\vec{a} = \hat{i} + \hat{j} \text{ (let)}$ $\therefore \vec{b}_1 \parallel \vec{a} \quad \vec{b}_1 = \frac{\vec{b} \cdot \vec{a}}{ \vec{a} } \cdot \hat{a} = \frac{3}{2} (\hat{i} + \hat{j})$ $\vec{b}_1 + \vec{b}_2 = \vec{b} \Rightarrow \vec{b}_2 = \frac{3}{2} \hat{i} + \frac{3}{2} \hat{j} + 4 \hat{k}$	Ⓑ	[4]



69.	$ f(x) + g(x) = \begin{cases} 7-2x & \text{if } x < 3 \\ 1 & \text{if } 3 < x < 4 \\ 2x-7 & \text{if } x > 4 \end{cases}$ $f(x) - g(x) = -1 \Rightarrow f(x) - g(x) = 1$ $\therefore \text{L. H. S} > 1$ $\Rightarrow x \in \mathbb{R} - [3, 4]$	Ⓓ	[4]
70.	$f(g(x)) = 3g(x) - 1$ <p>Range of $f(g(x)) = \mathbb{R} - \{2\}$ + codomain of $f(g(x))$</p> $\Rightarrow f(g(x)) \text{ is not onto.}$ $f(g(x_1)) = f(g(x_2)) \Rightarrow x_1 = x_2 = f(g(x)) \text{ is one-one.}$	Ⓒ	[4]
71.	<p>2401 .</p> $\bar{A} \times \bar{B} = 2(\bar{a} \times \bar{b})$ $= \bar{A} \times \bar{B} ^2 = 4(\bar{b} ^2 \bar{a} ^2 - (b \cdot a)^2)$ $= 4(2401 - (b \cdot a)^2)$ $= \bar{A} \times \bar{B} = 2(2401 - (b \cdot a)^2)^{1/2} \Rightarrow \lambda = 2401.$		[4]
72.	<p>0</p> $F(x) = \int_0^x f(t) dt \Rightarrow F'(x) = f(x).$ $I = \int_0^\pi (f'(x) + f(x)) \cos x dx = \int_0^\pi f'(x) \cos x dx + \int_0^\pi f(x) \cos x dx$ $J = \int_0^\pi f'(x) \cos x dx$ $= 4 - f(0) + \int_0^\pi \sin x f(x) dx$ $= 4 - f(0) + \int_0^\pi \pi \sin x F'(x) dx$ $= 4 - f(0) + 0 - \int_0^\pi \pi \sin x f(x) dx$ $\therefore \int_0^\pi f'(x) \cos x dx + \int_0^\pi \cos f(x) dx = 4 - f(0) = 4 - 4 = 0.$		[4]

73.	<p>4</p> <p>$S(Q \cup R) = S(Q) + S(R) - S(Q \cap R)$; $Q = \{101, 110, \dots, 992\}$</p> <p>$S(Q) = \frac{100}{2}(101 + 992) = 54650$</p> <p>Case 1: If $l = 2$ the $Q \cap R = Q \therefore S(Q \cup R) = S(Q)$ not possible as given sum = 109500</p> <p>Case 2: If $l \neq 2$, then $Q \cap R = Q$</p> <p>$\therefore S(Q \cup R) = S(Q) + S(R) = 109500$</p> <p>$\Rightarrow 54650 = \sum_{k=11}^{110} (9k + l) = 109500 \Rightarrow l = 4$</p>		[4]
74.	<p>2</p> <p>For equal roots: $(a - b)^2 - 4(1 - a - b) > 0$</p> <p>$\Rightarrow b^2 + b(4 - 2a) + a^2 + 4a - 4 > 0$</p> <p>$\Rightarrow (4 - 2a)^2 - 4(a^2 + 4a - 4) < 0$</p> <p>$\Rightarrow a > 1$.</p>		[4]
75.	<p>(256)</p> <p>$A^n = \begin{bmatrix} 2^n & 2^n - (-1)^n \\ 0 & (-1)^n \end{bmatrix}$</p> <p>$2A \cdot (\text{adj } 2A) = 2A I$</p> <p>$\Rightarrow A \cdot (\text{adj } 2A) = -2^2 \cdot I$</p> <p>Now, $A^{10} - \text{adj}(2A)^{10} = \frac{1}{ A ^{10}} A^{20} - 2^{20}I = 0$.</p> <p>$\therefore A^8 + A^{10} - \text{adj}(2A)^{10} = A ^8 + 0 = (-2)^8 = 256$.</p>		[4]