Monthly Progressive Test
Class: IX

## Subject: PCMB

## Solutions

## Physics

1. ©

$$
\frac{10-40}{6}=-5 \mathrm{~m} / \mathrm{s}^{2}
$$

2. (D)

If $u=0, s$ is directly proportional to $t^{2}$
$\frac{S_{1}}{S_{2}}=\left(\frac{1}{\sqrt{3}}\right)^{2}=\frac{1}{3}$
3. (D)
as velocity is constant, therefore acceleration is zero
4. (A)
$S=0+\left(\frac{a}{2}\right)(6-1)=9 \Rightarrow a=3.6 \mathrm{~m} / \mathrm{s}^{2}$
5. (B)

$$
h=\frac{5^{2}}{2 g} \Rightarrow 10 h=12.5 \mathrm{~m}
$$

6. ©
$h_{1}=\frac{1}{2} \times 10 \times t_{1}^{2} \Rightarrow 500=5 t_{1}^{2} \Rightarrow t_{1}=10 \mathrm{~s}$
Again, $500=330 \times t_{2} \Rightarrow t_{2}=1.5 \mathrm{~s}$ (nearly)
Therefore, total time $=10+1.5=11.5 \mathrm{~s}$
7. (B)

5 steps require 5 s , as 2 m progress requires 8 s , therefore 6 m progress requires $8 \times 3=24 \mathrm{~s}$. So total time required $=24+5=29 \mathrm{~s}$.
8. (A)

Yes
9. ©

Speed $=2 \pi r / t$
10. (B)
$\mathrm{m} / \mathrm{s}$ for both
11. ©
$0 \mathrm{~m} / \mathrm{s}$
12. (D)
cm/s
13. ©

Retardation
14. ©
$10 \mathrm{~m} / \mathrm{s}$
15. ©
$\mathrm{m} / \mathrm{s}^{2}$
16. (B)

As impulse $=F \cdot t=m v-m u$
17. (D)

Mole
18. (B)

As pressure $\times$ area $=$ Force
19. ©

At highest point, velocity is zero
20. ©

As speed cannot be negative
21. (A)

As $V^{2}=u^{2}+2 a s$, put $u=0 \mathrm{~m} / \mathrm{s}$
22. ©

As $s=\frac{u+v}{2} \times \mathrm{time}$
23. (A)

Apply $x=\frac{1}{2} a(2 t-1)$, as $u=0 ; t=3 \mathrm{~s}$
24. (A)
$\frac{\tan 60^{\circ}}{\tan 30^{\circ}}=3$
25. (B)

As ' $g$ ' is acting downward with constant magnitude. (Vector plotting)

## Chemistry

26. ©
$\mathrm{AB}_{2}$;
A has valency 2
$B$ has valency 1
$\therefore$ Formula is $\mathrm{AB}_{2}$
27. (B)

3 \& 4 ;
In $\mathrm{XCl}_{3}$, valency of ' X ' = 3
In $\mathrm{YCl}_{4}$, valency of ' Y ' $=4$
28. (D)
$\mathrm{CO}_{2}$ in solid state
Dry ice solid $\mathrm{CO}_{2}$
29. ©

Both A \& B;
Rate of evaporation depends on both temperature and surface area.
30. (B)

Boiling point;
Bydefinition boilingpoint is the temperature at whichliquid starts boiling at atmospheric pressure.
31. (A)

If a perfume bottle is opened in one corner of room, the smell can be felt after sometime in the opposite corner, then the particles of matter are constantly moving due to diffusion.
32. ©

Dalton's atomic theory
All matter is made up of very small particles which can not be further broken down. These particles are called atoms.
33. (D)
$\mathrm{M}_{2} \mathrm{O}_{3}$;
In $\mathrm{MCl}_{3}$, the valency of ' M ' is 3 .
So, the formula of metal oxide is $\mathrm{M}_{2} \mathrm{O}_{3}$ (since oxygen valence is 2)
34. ©

5;
In $\mathrm{A}_{2} \mathrm{O}_{5}$ the valency of A is 5 .
35. (D)
(A) is false and (R) is true fact.
36. ©

960 Cal ;
Heat $(Q)=$ mass $(m) \times$ Latent heat of melting $(L)$

$$
=12 \mathrm{~g} \times 80 \mathrm{Cal} / \mathrm{g}=960 \mathrm{Cal}
$$

37. (A)

Evaporation is called as surface phenomenon.
38. (B)

Boiling of a liquid takes place at a fixed temperature and normal atmospheric pressure.

[^0]39. (A)

Latent heat;
The amount of heat causes the change of state, but the temperature of a substance to remain constant is called latent heat.
40. (D)

Diffusion;
Movement of the particles from higher concentration to lower concentration is called diffusion. Here, incense stick when lit the aroma smoke diffuses in all parts of the room.
41. (D)

Nitrogen;
Carbon dioxide $\left(\mathrm{CO}_{2}\right)$ is a compound, water $\left(\mathrm{H}_{2} \mathrm{O}\right)$ is a compound. Air is a mixture of $\mathrm{N}_{2}, \mathrm{O}_{2}, \mathrm{CO}_{2}$ etc.
42. ©
$\mathrm{KMnO}_{4}$;
Chemical formula of Potassium permanganate is $\mathrm{KMnO}_{4}$.
43. (A)

Liquids and gases together is called fluids.
44. (B)

Arsenic;
Carbon is non-metal
Iron is metal
Sodium is metal
45. (B)

Rate of evaporation depends on surface area, temperature and humidity but it does not depend on material of the vessel.
46. ©

The chemical formula of Aluminium sulphate is $\mathrm{Al}_{2}\left(\mathrm{SO}_{4}\right)_{3}$.
Valency of Al = 3
Valency of $\mathrm{SO}_{\overline{4}}=2$
47. (D)

Balance equation is:
$2 \mathrm{~Pb}\left(\mathrm{NO}_{3}\right)_{2} \rightarrow 2 \mathrm{PbO}+4 \mathrm{NO}_{2}+\mathrm{O}_{2}$
So, $a=2, b=2, c=4$
48. (D)

States of matter are:
Solid, Liquid, Gas, Plasma, B.E.C. (Bose Einstein Condensate)
49. (B)

Sublimation: Conversion of solid into vapours is called sublimation.
Vapourisation: Conversion of liquid into gas.
Freezing: Conversion of liquid into solid.
50. (D)

## Mathematics

51. (D)

$$
\begin{aligned}
& x+y=2013, \quad \frac{1}{x}+\frac{1}{y}=2013 \\
& \Rightarrow \frac{y+x}{x y}=2013 \quad \Rightarrow \frac{2013}{x y}=2013 \Rightarrow x y=1
\end{aligned}
$$

52. (D)
$2 x-5 y+k=0$
$2(3)-5(-1)+k=0 \quad \Rightarrow k=-11$
53. ©
$2 x-3 y=7 ; \quad 4 x-6 y=20 \quad \Rightarrow 2 x-3 y=10$
$\frac{2}{2}=\frac{-3}{-3} \neq \frac{7}{10}$
$\therefore$ no solution
54. (D)
$2 x+3 y=6$. It passes through $(3,0)$ and $(0,2)$.
IV will be the graph.
55. (D)
$3 x-4 y=7 ; x+c y=13$
$\frac{3}{1}=\frac{-4}{c} \neq \frac{7}{13} \Rightarrow c=-\frac{4}{3}$
56. (A)

The line $x-7=0$ is parallel to $y$-axis.
57. ©

Let no. of chickens be $x$ and no. of goats be $y$.
$\therefore x+y=30 \Rightarrow x=30-y$
$2 x+4 y=84 \quad \Rightarrow x+2 y=42$
$\therefore 30-y+2 y=42 \quad \Rightarrow y=12$
$\therefore x=18$
$x: y=18: 12=3: 2$
58. (B)

$$
\frac{C}{5}=\frac{F-32}{9} \Rightarrow \frac{C}{5}=\frac{37-32}{9} \Rightarrow C=\frac{25^{\circ}}{9}
$$

59. (A)
$\frac{C}{5}=\frac{C-32}{9} \Rightarrow 9 C=5 C-160 \Rightarrow 4 C=-160 \Rightarrow C=-40$
60. (A)

$$
3 \times 4=a \times 3+7 \Rightarrow 3 a=5 \Rightarrow a=\frac{5}{3}
$$

61. (B)
$x+y=0$ is satisfied by $(2,-2),(0,0)$ and $(-3,3)$.
62. (B)
$a \neq 0, \quad \mathrm{~b} \neq 0$
63. (A)

$$
x+y=12, \quad x y=35
$$

$\therefore \frac{1}{x}+\frac{1}{y}=\frac{y+x}{x y}=\frac{12}{35}$
64. (A)
$\frac{4}{x}+5 y=7, \quad \frac{3}{x}+4 y=5$
$\left(\frac{1}{3},-1\right)$ satisfy both the equations.
65. (D)

$$
\begin{aligned}
& 1=a-b \Rightarrow a=1+b \\
& 5=a-\frac{b}{5} \Rightarrow 5=1+b-\frac{b}{5} \Rightarrow 4=\frac{4 b}{5} \Rightarrow b=5 \\
& \therefore a=6 \\
& \therefore a+b=6+5=11
\end{aligned}
$$

66. (A)

$$
\begin{aligned}
& x=\frac{1}{3-2 \sqrt{2}}=3+2 \sqrt{2} \\
& y=\frac{1}{3+2 \sqrt{2}}=3-2 \sqrt{2} \\
& x+y=6 ; \quad x y=1 \\
& \therefore x^{2}+y^{2}=(x+y)^{2}-2 x y=36-2=34
\end{aligned}
$$

67. (A)

$$
\begin{aligned}
& \sqrt{x}+\frac{1}{\sqrt{x}}=2 \Rightarrow x+\frac{1}{x}+2=4 \Rightarrow x+\frac{1}{x}=2 \Rightarrow x^{2}+1=2 x \Rightarrow(x-1)^{2}=0 \Rightarrow x=1 \\
& \therefore x^{8}+\frac{1}{x^{8}}=1+1=2
\end{aligned}
$$

68. (D)

Abscissa of all the points on $x$-axis is any number.
69. (A)
$P Q=5$ units
$\therefore$ Co-ordinates of $S$ is $(-3,3)$
70. (B)
$P(-5,3)$ lies in quadrant II.
71. (A)

$$
\begin{aligned}
& x+y=2 \quad \Rightarrow y=2-x \\
& \therefore x-y=1 \quad \Rightarrow x-(2-x)=1 \quad \Rightarrow x-2+x=1 \quad \Rightarrow 2 x=3 \quad \Rightarrow x=\frac{3}{2}
\end{aligned}
$$

72. (B)
$x-y=1$,
$x^{2}-y^{2}=3 \quad \Rightarrow(x+y)(x-y)=3 \quad \Rightarrow(x+y) \times 1=3 \quad \Rightarrow x+y=3$
73. (D)
$(4,0),(0,4),(2,2)$ satisfy the equation $x+y=4$
74. (B)
$x-y=3 \quad \therefore y=x-3$
75. ©
$x+y=3, x-y=1$
$\therefore x^{2}-y^{2}=(x+y)(x-y)=3 \times 1=3$

## Biology

76. (A)

Amitosis
77. (A)

Flemming
78. (A)

Two daughter cells are produced
79. (A)

M phase
80. (B)

## Diploid

Diploid mother cell produces haploid gametes through meiosis.
81. (B)

## Diploid

Mitosis is an equational division.
82. ©

S phase
Therefore called Synthetic phase.
83. ©

Nageli
84. (B)

Centrosome
85. (A)

Karyokinesis
86. (D)

Prophase $\rightarrow$ Metaphase $\rightarrow$ Anaphase $\rightarrow$ Telophase
87. ©

All living cells of the body, except those leading to the formation of gametes.
88. (B)

Equational division
Results in the formation of two daughter cells, identical to the mother cell.
89. (D)

Four
90. (D)

Both (A) and (B)
91. (A)

Cytoplasm

## 92. (A)

One
93. (B)

Chloroplast
94. (D)

Lysosome
In old and worn out cells, lysosomes burst to release its hydrolytic enzymes. These enzymes digest the whole cell.
95. ©

Protein synthesis
96. (B)

The nuclear region is covered by the nuclear membrane.
The nuclear region is naked in prokaryotic cells and lies freely in the cytoplasm.
97. ©

Producing energy during respiration.
98. (D)

No change will be observed in the size of the cell
In isotonic solutions, exosmosis = endosmosis. Hence no change is seen in shape and size of the cell.
99. ©

Are devoid of ribosomes.
100. (A)

Chromoplast
Contains pigments other than chlorophyll.


[^0]:    Techno India Group • DN-25

