

Monthly Progessive Test

Class: IX

Subject: PCMB



Time: 180 mins Full Marks: 200

Important Instructions:

- 1. The Test is of 180 mins duration and the Test Booklet contains 100 multiple choice questions of single correct option only. There are four sections with four subjects. You have to attempt all 100 questions (Candidates are advised to read all 100 questions). Questions 1 to 25 contain Physics, Questions 26 to 50 contain Chemistry, Questions 51 to 75 contain Mathematics, Questions 76 to 100 contain Biology.
- 2. Each question carries 2 marks. For each correct response, the candidate will get 2 marks. There is no negative mark for wrong response. The maximum mark is 200.
- 3. Use Blue / Black Ball point Pen only for writing particulars marking responses on Answer Sheet.
- 4. Rough work is to be done in the space provided for this purpose in the Test Booklet only.
- 5. On completion of the test, the candidate must handover the Answer Sheet to the invigilator before leaving the Room / Hall. The candidates are allowed to take away this Test Booklet with them.
- 6. The CODE for this Booklet is Off Line MPT0210052024.
- 7. The candidates should ensure that the Answer Sheet is not folded. Do not make any stray marks on the Answer Sheet. Do not write your UID No. anywhere else except in the specified space. Use of white fluid for correction is NOT permissible on the Answer Sheet. **Do not scrible or write on or beyond discrete bars of OMR Sheet at both sides**.
- 8. Each candidate must show on-demand his/her Registration document to the Invigilator.
- 9. No candidate, without special permission of the Centre Superintendent or Invigilator, would leave his/her seat.
- 10. Use of Electronic Calculator/Cellphone is prohibited.
- 11. The candidates are governed by all Rules and Regulations of the examination with regard to their conduct in the Examination Hall. All cases of unfair means will be dealt with as per Rules and Regulations of this examination.
- 12. No part of the Test Booklet and Answer Sheet shall be detached under any circumstances.
- 13. There is no scope for altering response mark in Answer Sheet.

Space For Rough Works



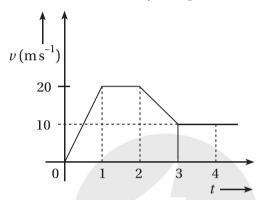
Physics

- 1. The ratio of magnitude of displacement to distance is always:
 - (A) less than 1

B greater than 1

© equal to 1

- less than or equal to 1
- **2.** The variation of the velocity of a particle moving along a straight line is illustrated in the graph given below. The distance covered by the particle in 3 seconds is (m).



A 20

B 35

© 40

- **D** 45
- **3.** A person travels along a straight road for the first half length with a constant speed v_1 and the second half length with a constant speed v_2 . The average speed V is
 - **A** $(v_1 + v_2)^{1/2}$
- **B** $\frac{2v_1v_2}{v_1+v_2}$
- \bigcirc $(v_1v_2)^{1/2}$
- **4.** A 120 m long train is moving to north at a speed of 9 m/s. A parrot flying towards south with a speed of 3 m/s crosses the train. The time taken by the parrot to cross the train would be
 - **(A)** 30 s

B 15 s

© 8 s

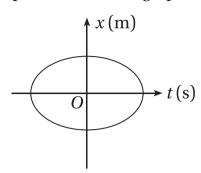
- **1**0 s
- **5.** How long will it take to stop a car travelling at a speed of $10 \, \text{m/s}$, if the uniform retardation. during braking is $1 \, \text{m/s}^2$?
 - **(A)** 100 s
- **B** 10 s

- © 0.25 s
- ① 0.01 s
- **6.** A particle starts moving from the position of rest under a constant acceleration. It travels a distance *x* in the first 10 sec and distance *y* in the next 10 sec, then
- © y = 3x
- \bigcirc y = 4x

- 7. A body falling under gravity moves with uniform
 - A Speed
- B Velocity
- © Momentum
- Acceleration

8.	A car starts from res		d achieves a veloci 2 m/s²	•	of 10 m/s in 5s.Wha 3 m/s ²		the acceleration? 4 m/s ²	
a	Starting from rest, a					_		
Э.	(a) 1 m/s		10 m/s		100 m/s	_	1000 m/s	
LO.	In the kinematic equations of motion, what does <i>v</i> represent?							
	A Initial velocity	lacksquare	Final velocity	©	Displacement	(Time	
L1.	When the object is a A straight line pa B A curve passing t C A straight line pa D Both of these	ssin hro	g through the origin		s time graph look?			
L2.	A man completes of then the displacement $\bigcirc P$	ent i	s				us of the circle is R ,	
10			$2\pi R$		$\pi R/2$			
LJ.	If an athlete takes tin						-	
	$\triangle r/t$	7	2r/t		$2\pi r/t$		$4\pi r/t$	
L4.	A man runs 10 m in							
	A 3 m/s	B	2 m/s	©	4 m/s	(D)	5 m/s	
L5.	36 km/hr =		OMINIA	C				
	A 1 m/s	B	5 m/s	©	10 m/s	(D)	0.1 m/s	
L6.	Dimension of pressu	ure	\times area \times (time) ²					
	A MLT	lacksquare	ML	©	MLT^2	(D)	ML^2T	
17.	Dimension of [lengt	h] -	[length]					
	(€ L ³	B	L^2	©	L	(D)	Dimensionless	
L8.	If strain is defined as	s ex	tension of length/o	rig	inal length, then di	ime	nsion of strain is	
	A MLT		ML^2T	_	MLT^2		Dimensionless	
L9.	The unit of impulse	is						
	Ns				kg m/s			
	© Dyne·s			(D)	All of these are co	rrec	et	
20.	In general, least cou	nt o	of vernier scale is					
	(A) 0.1 mm	lacksquare	0.01 mm	©	0.5 mm	(D)	0.05 mm	

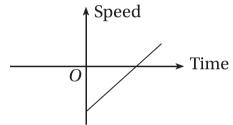
21. In one dimension motion, the position vs. time graph is shown below.



The graph is

- A Correct
- B Wrong
- © May be correct
- None of these

22.



The above graph is

- A Correct
- May be correct
- © Wrong
- None of these
- 23. The magnitude of velocity at highest point of vertical motion under gravity is
 - ② 2 m/s
- **B** 1 m/s
- $\bigcirc 0 \text{ m/s}$
- ① 10 m/s
- **24.** A particle is moving in straight line with uniform acceleration. Initial and final velocity of particle are 2 m/s and 4 m/s respectively. If the duration be 2 s, then displacement of particle is
 - **A** 6 m

B 3 m

- © 2.5 m
- ② 3.5 m
- **25.** In general, the stopping distance of a car after applying brake (symbols with usual notation) is

- $\odot \frac{2u^2}{a}$
- None of these

Chemistry

- **26.** Which of the following has the strongest interparticle force at the room temperature?
 - A Nitrogen
- B Mercury
- © Iron
- Chalk

- **27.** The boiling point of water on kelvin scale is
 - **(A)** 573 K
- **®** 273 K
- © 373 K
- **100 K**

28.	Fluids are (A) Liquids and gases (C) Liquids and solids	B	Solids and gases					
	© Liquids and solids	(D)	Only liquids					
29.	-	temperature at which a solid changes into liquid at atmospheric pressure is						
	A Melting point Boiling point	©	Condensation	Sublimation				
30.	The process for the change of a solid directly into its vapour is called—							
	A Evaporation B Ebullition	©	Condensation	Sublimation				
31.	Which of the following is not a chemical cl	han	ge?					
	A Passing of steam over red hot coke							
	Colour change of dilute nitric acid upon	n lo	ng standing					
	\bigcirc Absorption of moisture by P_2O_5							
	Absorption of moisture by CaCl ₂							
32.	Blackening of silverware is a							
	A Chemical change	lacksquare	Decomposition re	eaction				
	© Physical change	(D)	Evaporation proc	ess				
33.	Which one of the following set of phenomenon would increase on raising the temperature ?							
	Diffusion, evaporation, compression of gases							
	Evaporation, compression of gases, solubility							
	© Evaporation, diffusion, expansion of ga							
	© Evaporation, solubility, diffusion, comp		· ·					
34.	Which of the following causes the tempera it is undergoing a change in its state?	ıtur	e of a substance to	remain constant while				
	A Latent heat B Lattice energy	(C)	Loss of heat	None of these				
25		•	Loss of fiedt	e None of these				
<i>3</i> 5.	Gas has shape and volume		Definite in defini	4.0				
	Definite, definiteIndefinite, definite	BD	Definite, indefinite, indefinite, indefinite, indefinite, indefinite indefini					
00		U	maemite, maem	inte				
36.	The pressure exerted by a gas is due to			11 6.1				
	Inter particle collisions	B		alls of the container				
	© Gravity		Atmospheric pres					
37.	The volume of matter in 1 kg cotton is		_					
	A Smaller than B Greater than	(C)	Equal to	(D) Can not say				

38.	The type of motion t	that is present in gases	is						
	A Random		$^{f B}$	Linear (in a straig	ht li	ine)			
	© Vibratory		(D)	Circular					
39.	Evaporation is called	d as							
	Surface phenome	enon	B	Bulk phenomeno	n				
	© Both surface and	bulk phenomenon	(D)	Rare phenomeno	n				
40.	Boiling of a liquid ta	ıkes place at							
	(A) a fixed temperatu	a fixed temperature lower than its boiling point							
	B a fixed temperature	are and normal atmos	phe	ric pressure					
	© a fixed temperatu	are higher than its boil	ling	point					
	a fixed temperature	are and higher than at	mos	spheric pressure					
41.	The short form of an	element is known as							
	Compound		B	Molecule					
	© Symbol		(D)	Mixture					
42.	Which of the following	ing is an inert gas?							
	Arsenic	(B) Argon	©	Carbon	(D)	Iron			
43.		the element which m reactions is known as		or may not exist i	nde	pendently but take			
	Atom	Molecule	©	Symbol	(D)	Metal			
44.	Which of the following	ing is a metalloid?							
	A Carbon	Arsenic	©	Iron	(D)	Sodium			
45.	The symbol of cobal	t is							
	A Co	® Cb	©	Ct	(D)	C			
46.	The unit of pressure	of a gas is							
	A Pascal	B Kilogram	©	Metre	(D)	Cubic centimetre			
47.	. Ammonia and hydrogen chloride combine with each other to form ammonium chloride. This is an example of								
	Chemical change	e	B	Physical change					
	© Evaporation		(D)	Cracking					

- **48.** Which has the highest density?
 - (A) Gas

B Liquid

© Solid

- (D) All have same density
- 49. When sugar is added to water then
 - A Water evaporates

- Solution is formed
- © Sugar does not mix with water
- Water becomes solid
- **50.** On which factor the rate of evaporation does not depend
 - Surface area

Material of the vessel

© Temperature

Humidity

Mathematics

- **51.** Which among the options is one of the factors of $x^2 + \frac{x}{6} \frac{1}{6}$?
 - **(A)** 3x + 1
- © $x \frac{1}{5}$

- **52.** If $x \frac{1}{x} = 2$, then the value of $x^4 + \frac{1}{x^4}$ is
 - **A** 4

B 8

© 12

② 34

- **53.** If a + b + c = 0, then $\frac{a^2}{bc} + \frac{b^2}{ac} + \frac{c^2}{ab} =$
 - **(A)** 0

B 1

© −1

(D) 3

- **54.** If $a^{1/3} + b^{1/3} + c^{1/3} = 0$, then
 - **(A)** a + b + c = 0

B $(a+b+c)^3 = 27abc$

 \bigcirc a+b+c=3abc

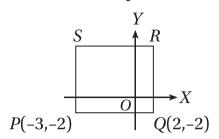
- **55.** Abscissa of all the points on *x*-axis is _____.
 - **(A)** 0

B 1

© 2

- any number
- **56.** The Cartesian system is named in honour of the mathematician _____.
 - A Leibnitz
- B Euclid
- © Laplace
- Rene Descartes

57. The given diagram is drawn on a cartesian plane



PQRS is a square. The coordinates of S are

- (-3, 3)
- **B** (3, -3)
- © (-3, -3)
- \bigcirc (-3, 2)

58. Which of the following expressions is a polynomial in one variable?

- (a) $x + \frac{2}{x} + 3$ (b) $3\sqrt{x} + \frac{2}{\sqrt{x}} + 5$ (c) $\sqrt{2}x^2 \sqrt{3}x + 6$ (d) $x^{10} + y^5 + 8$

59. $\sqrt{3}$ is a polynomial of degree

 $\bigcirc \frac{1}{2}$

© 1

(D) (I)

60. If p(x) = x + 4, then p(x) + p(-x) = ?

(A) 0

 \bigcirc 2x

(D) 8

61. If a + b + c = 12 and $a^2 + b^2 + c^2 = 50$, find the value of ab + bc + ca.

(A) 44

(B) 45

① 47

62. The point P(-5, 3) lies in

- (A) quadrant I
- © quadrant III
- quadrant IV

63. The point at which the two coordinate axes meet is called

- (A) the abscissa
- B the ordinate
- © the origin
- the quadrant

64. The perpendicular distance of the point A(7, 5) from y-axis is

- A 7 units
- **B** 5 units
- © 12 units
- © 2 units

65. The coordinates of two points are A(3, 4) and B(-2, 5), then (abscissa of A) – (abscissa of B) = ?

(A) 1

B −1

 \bigcirc 5

(D) -5

66. The value of $\sqrt[4]{\sqrt[3]{2^2}}$ is equal to

- $\triangle 2^{-1/6}$

© $2^{1/6}$

 $\bigcirc 2^6$

	[O]								
67.	$\sqrt[3]{2} \times \sqrt[4]{2} \times \sqrt[12]{32} = ?$								
	A 2		© $2\sqrt{2}$	① $4\sqrt{2}$					
68.	$\frac{(X^{a+b})^2(X^{b+c})^2(X^{c+a})^2(X^{c+a})^4}{(X^a.X^b.X^c)^4}$	$(2)^2$							
	♠ −1	B 0	© 1	None of these					
69.	The value of $\frac{(25)^{5/2}}{(16)^{3/4}}$	$\frac{(243)^{2/5}}{(8)^{5/3}}$ is							
			© $\frac{5625}{256}$	None of these					
70.	If $a = \frac{1}{3 - 2\sqrt{2}}$, $b = \frac{1}{3 - 2\sqrt{2}}$	$\frac{1}{+2\sqrt{2}}$ then the value o	of $a^2 + b^2$ is						
	(A) 34	B 35	© 36	D 37					
71.	If $x^2 + 2x = 45$, what	is the value of $x^4 + 4x^3$	$+4x^2 - 13$?						
	(A) 2013	B 1986	© 2012	③ 32					
72.	. If $x^3 - 5x^2 + 7$ is divided by $(x + 2)$, then the remainder is								
	A -21	B -20	© -17	D -25					
73.	If $a = 2^{\frac{1}{3}} = 2^{-\frac{1}{3}}$, then	find the value of $2a^3$ +	6a - 3						
	(A) 1	® −1	© 0	© 2					
74.	What is the quadrilateral that is formed by joining the points $(1, 1)$, $(2, 4)$, $(8, 4)$ and $(10, 1)$?								
	A triangle	B A square	© A rectangle	A trapezium					
75.	A straight line paral	lel to the x-axis has eq	uation						
	Biology								

- **76.** Hydrolytic enzymes are located in :
 - A Lysosomes
- B Ribosomes
- © Microsomes
- Mesosomes

- **77.** Tonoplast is a membrane surrounding the :
- B Nucleus
- © Vacuole
- Mitochondria

78.	Within the cell, the		f respiration or ox Nucleolus		tive phosphorylati Golgi bodies		is : Ribosomes
79.	Choose the correct option :				Power house - Golgi body Director of cell - Chloroplast		
80.	Example of uni-mer (A) Vacuole		nous cell organel Golgi body	_	: ER	(D)	All of these
81.	SER takes part in sy. (a) Lipids and steroi (b) Carbohydrate		sis of:	BD	Vitamins All of the above		
82.	Golgi apparatus is c		rned with : Secretion	©	ATP synthesis	(D)	RNA synthesis
83.	Grana & stroma oc A Ribosome		n— Chloroplast	©	Mitochondria	(D)	Golgi body
84.	Mitochondria were		observed by— Kolliker	©	Schwann	(D)	R. G. Harrison
85.	'Suicidal bag' is the ② ER	_	name for— Vacuole	©	Ribosome	(D)	Lysosome
86.	Kitchen of the cell is Mitochondria	_	ER	(C)	Chloroplast	(D)	Golgi apparatus
87.	The energy currency		cell is— AMP	©	АТР	(D)	СТР
88.	Ribosomes are cent Respiration	_		©	Protein synthesis	(Fat synthesis
89.	Which cell organelle Nucleus Ribosome	e is no	ot bounded by a r	$^{f B}$	nbrane? Lysosome ER		
90.	Cell organelle, com (a) Mitochondria (b) Ribosome	mon i	to prokaryotes an	B	ukaryotes is Golgi body Endoplasmic retic	culu	ım

		_	[
91.	Controlling centre of	of a o	cell is:						
	A Nucleus	B	Nucleolus	©	Chloroplast	(D)	Ribosome		
92.	The plasma membr	ane	of all cells is		<u> </u>				
				lacksquare	Semi permeable				
	© Permeable			(D)	Selectively perme	abl	e		
93.	Which one is not a part of nucleus?								
	A Chromatin	lacksquare	Nucleolus	©	Centrosome	(D)	Nucleoplasm		
94.	Which of the follow	ing	cells do not have a	a we	ll defined nucleus?)			
	♠ Egg cell	lacksquare	Neuron	©	Bacterial cell	(D)	All of the above		
95.	When a raisin is put in water, it swells up after sometime due to the process of:								
	Diffusion			lacksquare	Exosmosis				
	© Endosmosis			(D)	None of the above	е			
96.	The plastid that has	no j	pigment is						
	A Chloroplast	lacksquare	Chromoplast	©	Leucoplast	(D)	All of the above		
97. RER is rough because of attack					ned to it.				
	A ribosome		-	B	centrosome				
	© lysosome		m -	(D)	nuclear membrar	ıe			
98.	In a mature plant co	ell,	9		20				
	A vacuoles are abs	ent.	1/0		CX				
	(B) vacuoles are small and scattered all over the cytoplasm.								
	© vacuoles are located centrally and are small.								
	a single big vacu	ole (occupies almost t	ne er	ntire inner space of	f the	e cell.		
99.	The golgi apparatus	s is iı	nvolved in the for	nati	on of				
	A Ribosome			B	Centrosome				
	© Lysosome			(D)	Endoplasmic reti	culı	ım		
100.	. Transport of substances between the cytoplasm and nucleus occurs through the								
	♠ Endoplasmic ret	icul	um		Golgi body				
	© Mitochondria			(D)	Lysosome				

Space For Rough Works



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