

# **Monthly Progessive Test (Solution)**

Class: IX



## Subject: PCMB

Test Booklet No.: MPT02

Test Date: 1 0 0 5 2 0 2 4

**Physics** 

**1**. D

| distance  $| \ge |$  displacement |

**2**. D

$$\frac{1}{2} \times 1 \times 20 + 1 \times 20 + 1 \times 10 + \frac{1}{2} \times 1 \times 10 = 10 + 20 + 10 + 5 = 45$$

3. ®

$$=\frac{x+x}{\frac{x}{v_1}+\frac{x}{v_2}}=\frac{2v_1v_2}{v_1+v_2}$$

4. D

$$t = \frac{120}{9+3} = 10 \text{ s}$$

5. ®

$$t = \frac{u}{a} = \frac{10}{1} = 10$$
 s

6. ©

$$x = \frac{1}{2}a \times 100$$
  
$$x + y = \frac{1}{2} \cdot a \cdot 400 \implies x + y = 4x \text{ by solving, we get } y = 3x$$

7. D

'g' constant.

8. ®

 $10 = a \times 5$   $a \Rightarrow 2 \text{ m/s}^2$ 

9. ©

 $V = gt \implies V = 10 \times 10 = 100 \text{ m/s} \quad [\text{as } u = 0]$ [1]

10. ® V = final velocity 11. © **∧** *X* **→** t 0 12. D Displacement is zero. 13. ©  $\frac{\text{Distance}}{\text{time}} = \frac{2\pi r}{t}$ **14**. (A)  $\frac{10+11}{2+5} = 3 \text{ m/s}$ 15. © 36 km/hr = 10 m/s 16. ® As pressure × area = Force 17. © [L] - [L] = [L]18. D  $\frac{\Delta L}{L} \equiv \text{Dimensionless}$ 19. D impulse = force × time 20. A L.C = 0.1 mm21. B t = 0 s, x has two values 22. ©

Speed can't be negative

23. ©

as instaneous rest, v = 0

**24**. (A)

$$\frac{2+4}{2} \times 2 = 6 \text{ m}$$

25. ®

$$(0)^2 = u^2 - 2as \implies s = \left(\frac{u^2}{2a}\right)$$

## Chemistry

26. ©

Both iron and chalk are solid but the interparticle force of attraction is higher in iron than chalk.

27. ©

Boiling point of water is 100°C. So, in Kelvin scale the temperature is (100 + 273) = 373 K

28. **(**A)

The term fluid means the substance which has the ability to flow from one place to the other. So, liquid and gas are the fluids

## 29. ®

**Boiling point :** It is the temperature at which one solid turns into liquid when pressure is one atmosphere

## 30. D

Sublimation : The process in which one solid turns into vapour when heat is supplied.

## 31. D

When solid calcium chloride absorbs water then it becomes soluble in water but the chemical property does not change.

## 32. **(**A)

It is a chemical change and related equations are given below

 $4Ag + O_2 \longrightarrow 2Ag_2O$ 

 $Ag_2O + H_2S \longrightarrow Ag_2S + H_2O$ 

## 33. ©

On increasing temperature, the rates of evaporation, diffusion, and expansion of gases increase sharply.

#### 34. ®

**Lattice energy :** It is the energy needed for the change of state of 1 mole of substance at a constant temperature

#### 35. D

Gases do not have definite shape and volume

#### 36. ®

Gas molecules move randomly inside the container and the perpendicular force applied by them on the inner wall of the container is the measure of pressure

#### 37. ®

Density of 1 kg sugar is higher than 1 kg cotton. Now, density =  $\frac{\text{mass}}{\text{volume}}$ . So, volume of 1 kg cotton is greater than that of 1 kg sugar.

#### 38. ®

When gas molecules move freely then they obey the linear motion

#### 39. **(**

Evaporation occurs from surface not from bulk

#### 40. ®

Boiling occurs at a particular temperature and at normal atmospheric pressure

#### 41. ©

Symbol

#### 42. ®

Argon is an inert gas. Other inert gases are helium, neon, krypton, xenon, radon

#### 43. **(A)**

Atom

#### 44. ®

Arsenic

#### 45. **(A)**

Со

[5]

## 46. 🕲

 $Pascal = Newton / meter^2$ 

## 47. 🕲

 $NH_3 + HCl \longrightarrow NH_4Cl$  it is a chemical change

## 48. ©

Solid

## 49. ®

Sugar is highly soluuble in water hence a solution is formed as no chemical change is going on here.

#### 50. ®

Material of the ccontainer is not a factor of evaporation of a liquid.

• Mathematics •
51. ®
$x^2 + \frac{x}{6} - \frac{1}{6}$
$=\frac{1}{6}(6x^{2}+x-1)=\frac{1}{6}(6x^{2}+3x-2x-1)$
$=\frac{1}{6}\{3x(2x+1)-1(2x+1)\}$
$=\frac{1}{6}(2x+1)(3x-1)$
52. <sup>(D)</sup>
$x - \frac{1}{x} = 2 \Longrightarrow x^2 + \frac{1}{x^2} = 6$
$\Rightarrow x^4 + \frac{1}{x^4} = 34$
53. D
$\frac{a^2}{bc} + \frac{b^2}{ac} + \frac{c^2}{ab} = \frac{a^3 + b^3 + c^3}{abc} = \frac{3abc}{abc} = 3$
54. ®
$a^{\frac{1}{3}} + b^{\frac{1}{3}} + c^{\frac{1}{3}} = 0$
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$$\therefore (a^{\frac{1}{3}})^3 + (b^{\frac{1}{3}})^3 + (c^{\frac{1}{3}})^3 = 3a^{\frac{1}{3}} b^{\frac{1}{3}} c^{\frac{1}{3}}$$
$$\Rightarrow a + b + c = 3(abc)^{\frac{1}{3}}$$
$$\Rightarrow (a + b + c)^3 = 27abc$$

#### 55. D

Any number

#### 56. D

**Rene Descartes** 

## **57**. (A)

PQ = 5 units, SP = 5 units

 $\therefore$  Co-ordinates of *S* is (-3, 3)

## 58. ©

 $\sqrt{2}x^2 - \sqrt{3}x + 6$ 

## 59. D

$$5\sqrt{3}$$
 is constant polynomial. So, degree = 0

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## 60. D

$$p(x) + p(-x) = x + 4 - x + 4 = 8$$

61. D

$$a + b + c = 12, a^{2} + b^{2} + c^{2} = 50$$
  

$$(a + b + c)^{2} = a^{2} + b^{2} + c^{2} + 2(ab + bc + ca)$$
  

$$\Rightarrow (12)^{2} = 50 + 2(ab + bc + ca)$$
  

$$\Rightarrow 144 - 50 = 2(ab + bc + ca)$$
  

$$\Rightarrow 94 = 2(ab + bc + ca)$$
  

$$\Rightarrow ab + bc + ca = 47$$

## 62. ®

P(-5, 3) lies in quadrant II

## 63. ©

The origin

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64.	A
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Distance of the point A(7, 5) from y-axis = 7 units

#### 65. ©

Abscissa of A — abscissa of B = 3 - (-2) = 5

#### 66. ©

$$\sqrt[4]{\sqrt[3]{2^2}} = (4^{\frac{1}{3}})^{\frac{1}{4}} = (4)^{\frac{1}{12}} = 2^{\frac{1}{6}}$$

67. **(**A)

$$\sqrt[3]{2} \times \sqrt[4]{2} \times \sqrt[1]{32}$$
$$= 2^{\frac{1}{3}} \times 2^{\frac{1}{4}} \times (32)^{\frac{1}{12}} = 2^{\frac{1}{3} + \frac{1}{4} + \frac{5}{12}} = 2^{\frac{12}{12}} = 2$$

68. ©

$$\frac{(x^{2a+2b} \times x^{2b+2c} \times x^{2c+2a})}{x^{4a} \times x^{4b} \times x^{4c}} = \frac{x^{4a+4b+4c}}{x^{4a+4b+4c}} = 1$$

69. D

$$\frac{(25)^{\frac{5}{2}} \times (243)^{\frac{2}{5}}}{(16)^{\frac{3}{4}} \times (8)^{\frac{5}{3}}} = \frac{(5)^5 \times (3)^2}{(2)^3 \times (2)^5} = \frac{28125}{256}$$

70. (A)

$$a = \frac{1}{3 - 2\sqrt{2}} = 3 + 2\sqrt{2}, b = \frac{1}{3 + 2\sqrt{2}} = 3 - 2\sqrt{2}$$
$$a^{2} + b^{2} = (3 + 2\sqrt{2})^{2} + (3 - 2\sqrt{2})^{2}$$
$$= 2(9 + 8) = 34$$

71. ©

$$(x^{2} + 2x)^{2} = (45)^{2}$$
  
 $\Rightarrow x^{4} + 4x^{3} + 4x^{2} = 2025$   
 $\Rightarrow x^{4} + 4x^{3} + 4x^{2} - 13 = 2012$ 

#### 72. **(**A)

$$p(x) = x^{3} - 5x^{2} + 7$$
$$p(-2) = -8 - 20 + 7 = -21$$

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## 73. ©

$$a = 2^{\frac{1}{3}} - 2^{-\frac{1}{3}}$$
  

$$\Rightarrow a^{3} = 2 - 2^{-1} - 3 \times 2^{\frac{1}{3}} \times 2^{-\frac{1}{3}} \times (2^{\frac{1}{3}} - 2^{-\frac{1}{3}})$$
  

$$= 2 - \frac{1}{2} - 3a = \frac{3}{2} - 3a$$
  

$$\Rightarrow 2a^{3} = 3 - 6a \Rightarrow 2a^{3} + 6a - 3 = 0$$

#### 74. D

A trapezium

#### 75. ®

A line parallel to the *x*-axis is given by y = a

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## Biology

## 76. 🕲

Lysosomes Helps in intracellular digestion.

#### 77. ©

Vacuole

#### 78. **(**A)

Mitochondria

#### 79. 🕲

Membrane biogenesis - ER

RER produces proteins, while SER produces lipids. Both of these are required for forming new membranes.

#### 80. D

All of the above

#### 81. (A)

Lipids and steroids

#### 82. ®

Secretion

## 83. ®

Chloroplast

## 84. ®

Kolliker

## 85. D

Lysosome

In case of old or worn out cells, the lysosome bursts to release its enzymes, which digests the entire cell.

## 86. ©

Chloroplast Site of photosynthesis.

## 87. ©

ATP

88. ©

Protein synthesis

## 89. ©

Ribosome

## 90. ©

Ribosome

## 91. 🕲

Nucleus

Regulates all the activities of a cell.

## 92. D

Selectively permeable

It allows only certain substances to enter or exit a cell.

## 93. ©

Centrosome It is a distinct organelle.

#### 94. ©

Bacterial cell



[10]

Its a prokaryotic cell.

## 95. ©

Endosmosis

Water being hypotonic, enters the cell through the process of osmosis.

## 96. ©

Leucoplast Its function is to store food, hence does not require pigments.

## 97. 🕲

Ribosome

## 98. D

A single big vacuole occupies almost the entire inner space of the cell

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## 99. ©

Lysosome

## 100. (A)

Endoplasmic reticulum